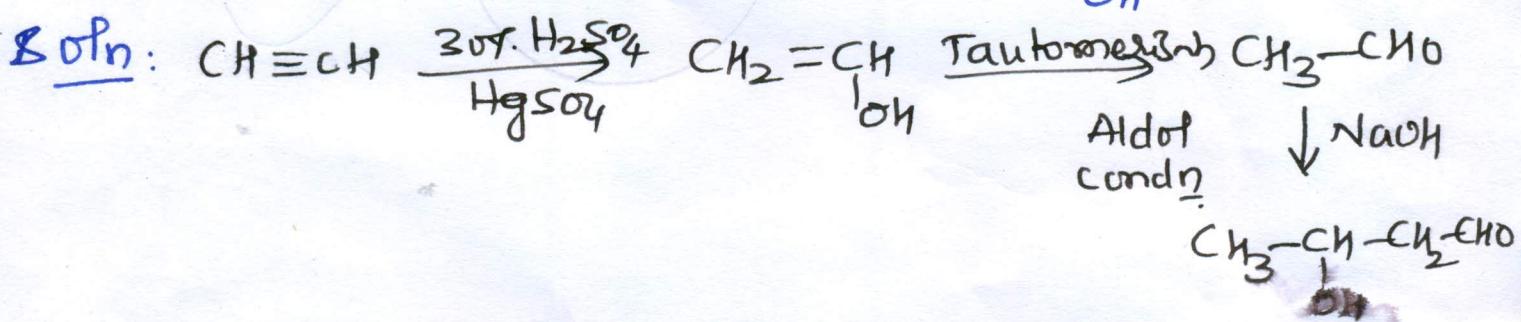
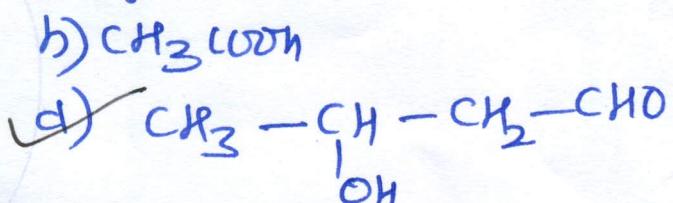


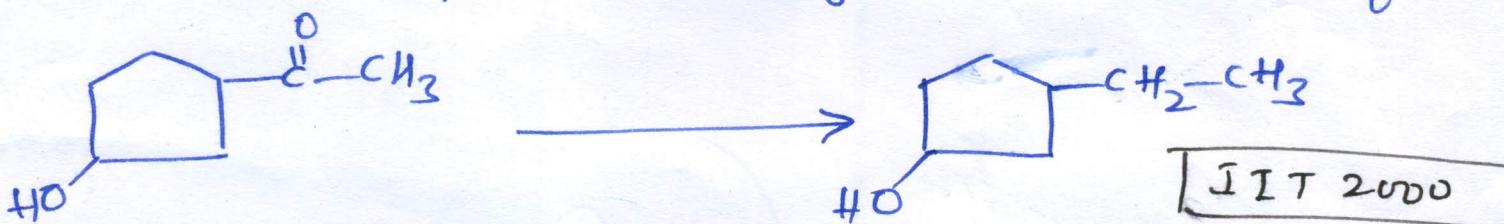
## Aldehydes & Ketones

① Predict the product 'B' in the sequence of reaction  $\text{CH} \equiv \text{CH} \xrightarrow[\text{HgSO}_4]{\text{30\% H}_2\text{SO}_4} \text{A} \xrightarrow{\text{NaOH}} \text{B}$

- a)  $\text{CH}_3\text{COONa}$
- b)  $\text{CH}_3\text{COOH}$
- c)  $\text{CH}_3\text{CHO}$



② The appropriate reagent for the transformation



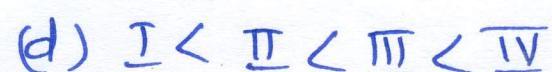
- a)  $\text{Zn-Hg, HCl}$
- b)  ~~$\text{NH}_2 - \text{NH}_2, \text{OH}^-$~~
- c)  $\text{H}_2, \text{Ni}$
- d)  $\text{NaBH}_4$

Soln

Both  $\text{Zn-Hg/HCl}$  &  $\text{NH}_2 - \text{NH}_2/\text{OH}^-$  can reduce  $\text{CH}_3 - \text{CO}-$  to  $\text{CH}_2 - \text{CH}_2 -$ , But  $\text{HCl}$  will react with  $-\text{OH}$  group

$\therefore \text{NH}_2 - \text{NH}_2/\text{OH}^-$  is more effective

Arrange the following compds in increasing order of their reactivity in nucleophilic addition reactions. Ethanal (I), Propanal (II), propanone (III), Butanone (IV).



Soln Reactivity of carbonyl ( $C=O$ ) gp. decreases with size of alkyl groups & no. of alkyl grps.  
 $\text{CH}_3\text{CHO}$  (I),  $\text{CH}_3\text{CH}_2-\text{CHO}$  (II),  $\text{CH}_3\text{COCH}_3$  (III)  $\text{CH}_3\overset{\oplus}{\text{C}}-\text{CH}_2-\text{CH}_3$  (IV)

Butanone < Propanone < Propanal < Ethanal

(A) consider the reaction:

AIPMT main  
2012



What sort of reaction is it?

- a) Electrophilic addition - Elimination reacn.
- b) Free radical addition - Elimination reacn.
- c) Electrophilic substitution - Elimination reacn.
- d) Nucleophilic addition - Elimination reacn.

